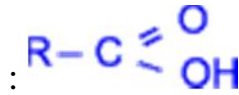


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COOH.



$nH_{2n}O_2$ ). :  $nH_{2n+1}COOH$  (

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		( )	$t_{\text{m.p.}}$ °C	$t_{\text{b.p.}}$ °C	$\rho$ / $g \cdot cm^{-3}$	$d_4^{25}$ ( /100 $H_2O$ ; 25°C)
		HCOOH	8,3	100,5	1,22	$\infty$
		CH <sub>3</sub> COOH	16,8	118	1,05	$\infty$
		CH <sub>3</sub> CH <sub>2</sub> COOH	-21	141	0,99	$\infty$
		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> COOH	-6	164	0,96	$\infty$
		CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> COOH	-34	187	0,94	4,97

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: 17 35 , 15 31 ;  
: 17 33 , 17 31 ,  
" 17 29 "

(C<sub>1</sub>-C<sub>4</sub>)

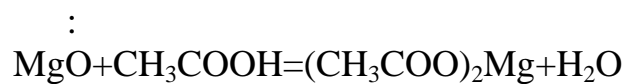
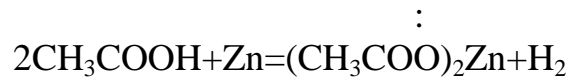
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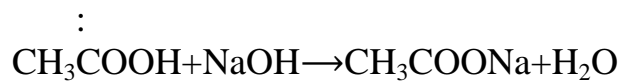


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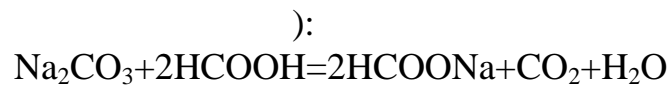
2.



3.



4.



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