

()

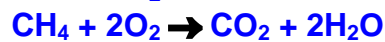
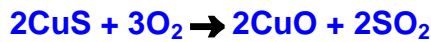
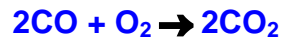
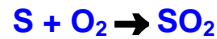
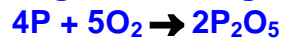
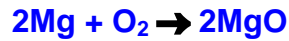
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: Na₂O; MgO; CuO

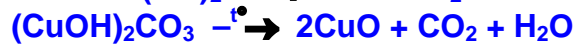
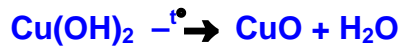
(): ZnO; Al₂O₃; Cr₂O₃; SnO₂

: SO₂; SO₃; P₂O₅;

1.



2.



1.	
$\text{Na}_2\text{O} + \text{H}_2\text{O} \rightarrow 2\text{NaOH}$ $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$ <p>!</p>	$\text{SO}_3 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_4$ $\text{P}_2\text{O}_5 + 3\text{H}_2\text{O} \rightarrow 2\text{H}_3\text{PO}_4$ <p>!</p>
2.	
$\text{MgO} + \text{H}_2\text{SO}_4 \xrightarrow{-t^\circ} \text{MgSO}_4 + \text{H}_2\text{O}$ $\text{CuO} + 2\text{HCl} \xrightarrow{-t^\circ} \text{CuCl}_2 + \text{H}_2\text{O}$	$\text{CO}_2 + \text{Ba}(\text{OH})_2 \rightarrow \text{BaCO}_3 + \text{H}_2\text{O}$ $\text{SO}_2 + 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_3 + \text{H}_2\text{O}$
3.	
$\text{Na}_2\text{O} + \text{CO}_2 \rightarrow \text{Na}_2\text{CO}_3$	
4.	
$3\text{CuO} + 2\text{NH}_3 \rightarrow 3\text{Cu} + \text{N}_2 + 3\text{H}_2\text{O}$ $\text{P}_2\text{O}_5 + 5\text{C} \rightarrow 2\text{P} + 5\text{CO}$	