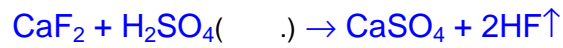
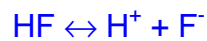


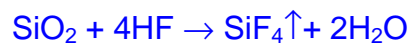
$t^{\circ} = -83,5^{\circ}\text{C}$ ;  $t^{\circ} = 19,5^{\circ}\text{C}$ ;



1) HF - (aq):

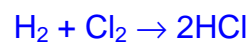


2) SiO<sub>2</sub> - (s):



400).  $t^{\circ} = -114^{\circ}\text{C}$ ,  $t^{\circ} = -85^{\circ}$ . (1 :

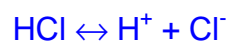
1) H<sub>2</sub> (g):



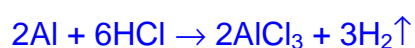
2) NaCl (s):

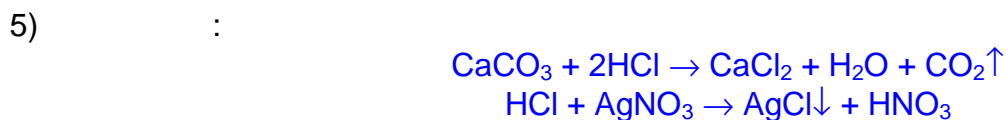
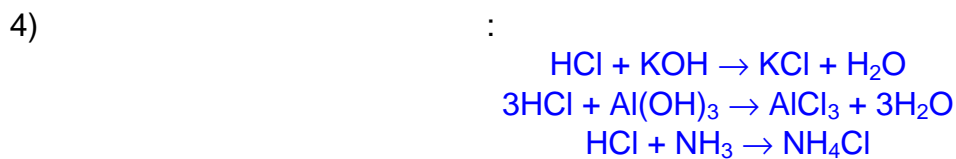


1) HCl - (aq):

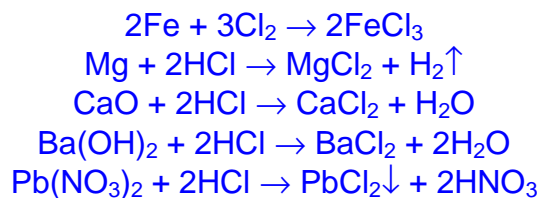


2) Al - (s):

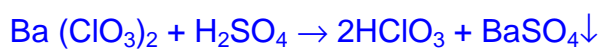
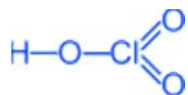
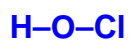


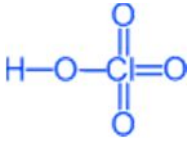


Cl<sup>-</sup>



).

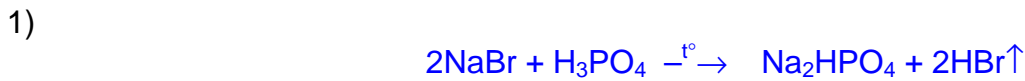




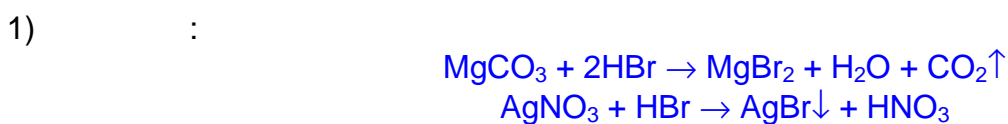
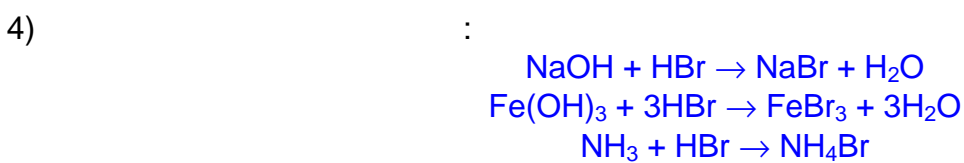
,  $t^\circ$  . = 25°C,  $t^\circ$  . = -101°C.



, ;  $t^\circ$  . = -67° ;  $t^\circ$  . = -87° .

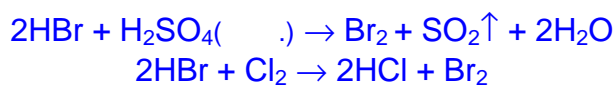


-  
HCl:



Br<sup>-</sup>

2) HBr -

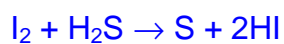


HBr<sup>+5</sup>O<sub>3</sub>.

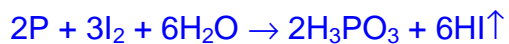
HBr<sup>+1</sup>O

, t° = -35° ; t° = -51° .

1)



2)

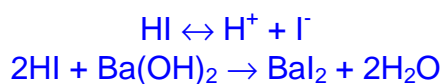


1)

HI

-

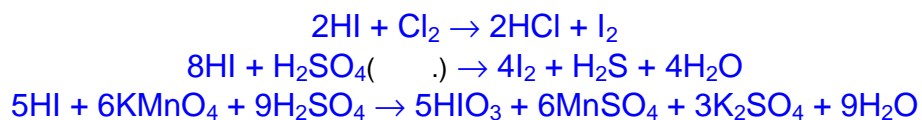
:



( HI - HCl HBr)

2) HI -

:



3)

I<sup>-</sup>

:

